

**SYNTHESIS, SPECTRAL STUDIES AND ANTI-BACTERIAL ACTIVITY OF 8-SUBSTITUTED-4-(2,4-DIHYDROXYPHENYL)-2-(4-CHLOROPHENYL)-2,5-DIHYDRO-1,5-BENZOTHIAZEPINES**

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Corresponding author e-mail:** lect.pernajain@gmail.com*Received on: 03-10-2015; Revised on: 19-11-2015; Accepted on: 21-12-2015ABSTRACT**

A new series of 1,5-benzothiazepine compounds have been synthesized by the reaction between equimolar quantities of 5-substituted-2-aminothiophenol and 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone, in the presence of absolute alcohol saturated with dry HCl gas. The reaction progress is monitored by TLC. Structure of compound is ascertained by Spectral and by elemental analysis.

Keywords: 1,5-benzothiazepine ; 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone ; anti-bacterial activity; 5-substituted-2-aminothiophenol.

INTRODUCTION

1,5-benzothiazepine scaffold is extremely versatile in a large no. of drugs therefore it attracted chemists in the field of drugs and pharmaceutical research. Benzothiazepine compounds are widely used as anticonvulsant¹, anti-spasmodic², neurolaptic³, antidepressive⁴, antibacterial⁵, anti-ulcer⁶ as well as anti-inflammatory⁷ agents. The I generation drug Diltiazem⁸ having 1,5-benzothiazepine nucleus with 4-methoxy group at position 2, acetoxy at position 3, oxo at 4. The prodigious applications of diltiazem such as blood platelet aggregation inhibitor⁹, anti thrombotic¹⁰, anti ischemic¹¹, anti arrhythmic¹², Calcium channel modulator¹³, anti-hypertensive¹⁴ etc have captivate the chemists to synthesize more improved version of it. Thus endeavour of researchers had launched Clentiazem¹⁵ in which Cl is present at position 8 and an isopropyl at position 5. Clentiazem exhibits more curative uses as anti-coagulant⁹, anti-arteriosclerotic¹⁶ etc. From short summary of literature it was concluded that inclusion of chlorine as chlorophenyl substituent in benzothiazepine nucleus intensify the biological

activities such as vasodepressant¹⁷, analgesic¹⁸ etc. This biological profile of 1,5-benzothiazepine incited us to synthesize new benzothiazepine derivatives containing chlorophenyl groups. The work has been done and presented in this communication.

MATERIALS AND METHODS

The m.p. of all the synthesized compounds were determined in open capillary tubes and are precise. The reaction progress is monitored by TLC on silica gel G using solvent system benzene: ethanol; ammonia(7:2:1). The IR spectra were taken in KBr pellets using Perkin-Elmer RX1 FT IR spectrometer. The ¹H NMR spectra were recorded on Bruker Avance 400 (FT NMR) using CDCl₃ as solvent. The mass spectra were recorded on JMS-T100LC, Accu TOF (DARTMS) spectrometer. The Elemental and Spectral analysis have executed at CDRI, lucknow.

Synthesis of 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone: 2,4-dihydroxy acetophenone

(0.01 mol, 1.51 g) and p-chlorobenzaldehyde (0.01 mol., 1.40 gm) were dissolved in 20 ml ice cold ethanol in round bottom flask placed in ice bath. To this 50 % NaOH solution was added with vigorous stirring for about 30 minutes. The reaction mixture was kept overnight then diluted with ice cold water and neutralized with dilute HCl. The creamy yellow crude product was obtained which was washed with cold water and dried. The yellow crystals of 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone were obtained. (m.p. 138⁰C lit. 134-137⁰C, yield 89%).

8-methoxy-4-(2,4-dihydroxyphenyl)-2-(4-chlorophenyl)-2,5-dihydro-1,5-benzothiazepine:

2-amino-5-methoxybenzothiol (0.001mol, 0.139gm) and 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone (.001 mol, 0.274 gm) were dissolved in 50 ml dry ethanol and saturated with dry HCl gas. The reaction mixture was refluxed for about five hours when colour change from pale yellow to deep red. The reaction mixture was reduced under diminished pressure to obtain crude solid which was crystallized by ethanol to give red brown coloured crystals of 8-methoxy-4-(2,4-dihydroxyphenyl)-2-(4-chlorophenyl) -2,5-dihydro-1,5-benzothiazepine 5d. (m.p 135⁰, yield 72%). On the above pattern all the compounds were prepared. The analytical and spectral data of 5a to f are given in table I & II respectively.

RESULTS AND DISCUSSIONS

To achieve the synthesis of targeted compound, the methodology is as follows;

Step-I Synthesis of 1-(2,4-dihydroxyphenyl)-3-(4-chlorophenyl)-2-propenone (3) is carried out by the condensation of 2,4-dihydroxyacetophenone (1) and 4-chlorobenzaldehyde (2). (Scheme I)

Step-II An equimolar quantity of chalcone and 5-substituted-2-aminothiophenol was refluxed in dry ethanol saturated with dry HCl gas. The red brown coloured crystals of 8-substituted-4-(2,4-dihydroxyphenyl)-2-(4-chlorophenyl)-2,5-dihydro-1,5-benzothiazepines were separated out and crystallized with ethanol. The structure of the final products was determined by IR, H¹NMR, Mass Spectra and Elemental Analysis.

The reaction between 5-substituted-2-aminothiophenol and chalcone initialise by nucleophilic attack of the lone pair of sulphhydryl electrons on the β-carbon of chalcone . It gives ketoamine which tautomerizes to the inolic form which on dehydration and cyclisation yields final product 8-substituted-4-(2,4-dihydroxyphenyl)-2-(4-chlorophenyl)-2,5-dihydro-1,5-benzothiazepine(5a-f).

Spectral Analysis: The absence of absorption band in the region 1685-1645^{cm-1} confirmed the absence of C=O group. However broad absorption around 3145-3130^{cm-1} was obtained which may due to secondary amino group. Peak near 3400-3300 ^{cm-1} confirms the presence of OH group. Band near 790-775 ^{cm-1} ascribed to C-Cl stretching. The H¹-NMR spectra of all final product (5a-f) showed one proton doublet at δ=6.45-7.35 (d, 1H, J=7 Hz) which may be assigned to C₂-H and other doublet around δ= 7.05-8.20 (d, 1H, J=7 Hz) integrating for 1 proton may be ascribed to C₃-H. The broad absorption signal in the region 3.85-4.30 (br, 1H) may be due to NH. The multiplet at around 6.52 to 8.15 (m,10H) may be assigned to aromatic ring. The absorption at 11.5-12.4 as a singlet may be assigned to OH group. In the spectra of 5d peak at δ=3.54 as a singlet may be assigned to 3 protons of OCH₃ group. The peak at δ 2.45 as a singlet integrated for 3 protons indicates the presence of CH₃ group in 5f. In spectra of compound 5e, the signal at 3.85 as quartet integrated for 3H and triplet at δ 1.90 integrated for 2H with same J=6 Hz may be assigned to OC₂H₅ group. The mass spectra of compounds 5c showed molecular ion peaks [M]⁺ and [M+2]⁺ peak at 440.4 and 442.4 correspond to the molecular mass of the product. Both the peaks were found to be nearly of equal intensity which confirms the presence of Bromine in 5c.

Anti-bacterial activity: Compound 5a-5f were investigated for their invitro anti-bacterial activity against E.coli and Ps. Aeruginosa using paper disc method¹⁹. Ampicillin and amikacin was used as a standard drug for E.coli and pseudomonas aeruginosa respectively. The concentration of compound used was 100 µg/disc. The zone of inhibition for all compounds was measured after 24 hours of incubation at optimum temperature.

CONCLUSION

The compounds 5a, b, c, d, f showed anti-bacterial activity almost equal to the reference drug Amikacin against Ps.Aeruginosa and compounds 5a, b, d, e, f showed anti-bacterial activity almost equal to the reference drug Ampicillin against E.Coli..

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Table-1
Characteristic data of compound 5a-5f

Compd	X	m.p °C	Yield (%)	R _f	Mol.formula (mol.wt.)	Elemental analysis Calculated(found)%		
						C	H	N
5a	F	132	82	0.72	C ₂₁ H ₁₅ O ₂ SNF (379.5)	66.40 (65.98)	3.95 (3.23)	3.68 (3.56)
5b	Cl	157	85	0.74	C ₂₁ H ₁₅ O ₂ SNCl (396)	63.63 (62.45)	3.78 (3.67)	3.53 (3.56)
5c	Br	116	79	0.65	C ₂₁ H ₁₅ O ₂ SNBr (440.5)	57.20 (56.65)	3.40 (3.56)	3.17 (3.12)
5d	OCH ₃	135	72	0.68	C ₂₂ H ₁₈ O ₃ SN (391.5)	67.43 (66.76)	4.59 (4.23)	3.57 (3.25)
5e	OC ₂ H ₅	145	69	0.64	C ₂₃ H ₂₀ O ₃ SN (405.5)	68.06 (67.920)	4.93 (4.46)	3.45 (3.23)
5f	CH ₃	139	78	0.71	C ₂₂ H ₁₈ O ₂ SN (375.5)	70.30 (68.93)	4.79 (3.94)	3.72 (3.62)

Table-2

Comp. No.	IR	¹ H-NMR				
	N-H	N-H (br, 1H)	C ₂ -H (d, 1H, J7)	C ₃ -H (d, 1H, J7)	O-H (s,1H)	Ar Protons (m, 10H)
5a	3135	3.95	6.45	7.05	11.4	6.54-7.90
5b	3140	3.98	6.99	7.10	11.9	7.20-7.85
5c	3134	4.04	7.30	7.80	12.3	7.10-8.10
5d	3136	4.02	7.35	8.10	12.1	7.29-7.99
5e	3138	4.02	6.89	7.10	11.7	6.99-7.86
5f	3141	4.25	7.15	7.12	11.8	7.11-7.99

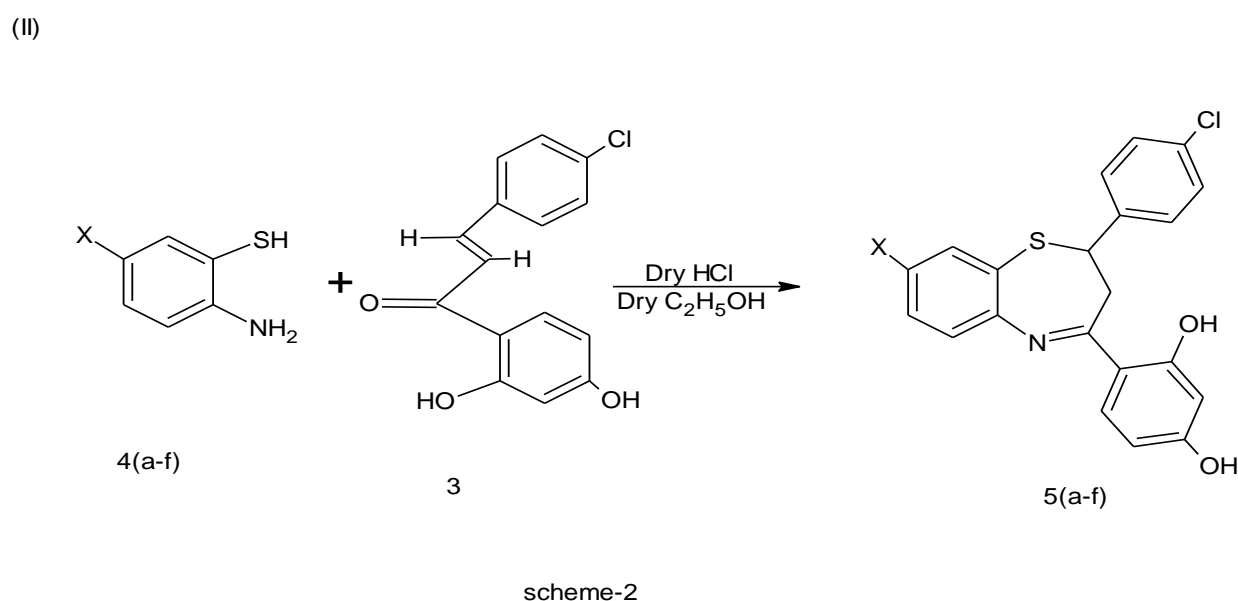
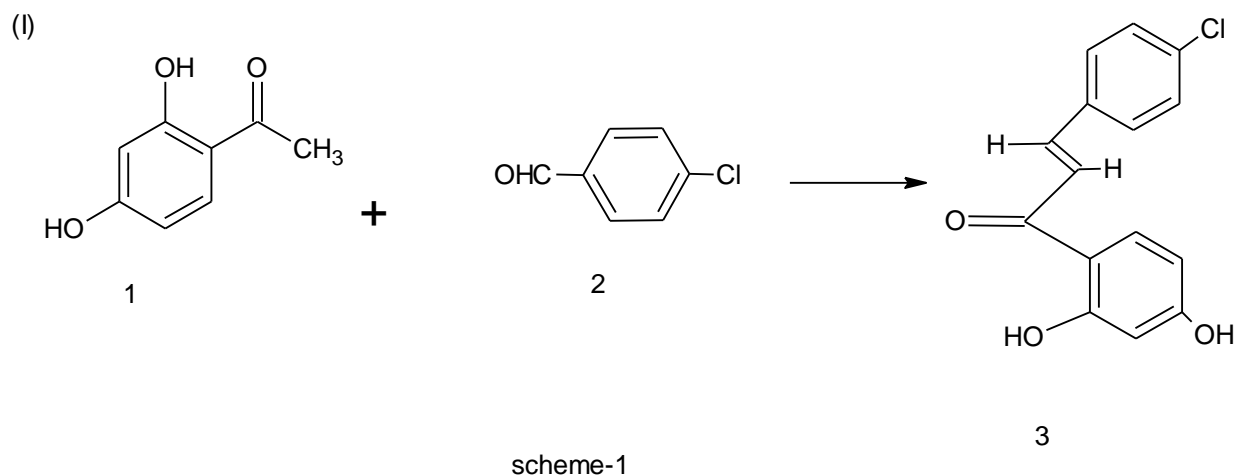
Spectral data of compd 5a-f

Table -3
Antibacterial studies of compds 5a-f

Compound	Zone of inhibition(mm)	
	E.coli	Ps.aeruginosa
5a	15	19
5b	16	18
5c	14	15
5d	13	20
5e	09	19
5f	14	17

Zone of inhibition of ampicillin= 18mm

Zone of inhibition of amikacin = 21mm



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