

Marmacy nternational Mournal of Pharmacy

Journal Homepage: http://www.pharmascholars.com

# **Original Article**

# **CODEN: IJPNL6**

# A STABILITY INDICATING METHOD FOR ESTIMATION OF TAPENTADOL IN BULK AND IN FORMULATIONS

Renu Chadha <sup>a,\*</sup>, Alka Bali<sup>a</sup>, Gulshan Bansal<sup>b</sup>

<sup>a</sup> University Institute of Pharmaceutical Sciences, UGC Center of Advanced Study, Panjab University, Chandigarh,160014, India.

<sup>b</sup> Department of Pharmaceutical Sciences and Drug Research, Punjabi University, Patiala 147002, Punjab, India.

# \*Corresponding author e-mail: renukchadha@rediffmail.com

Received on: 27-02-2016; Revised on: 16-03-2016; Accepted on: 28-03-2016

#### ABSTRACT

An isocratic stability-indicating reversed phase liquid chromatography (RP-HPLC-UV) method for quantitative determination of tapentadol HCl has been developed and validated as per the ICH guidelines. Tapentadol HCl and the tablet formulation were subjected to forced decomposition conditions of hydrolysis, oxidation, photolysis and thermal stress, as per ICH guidelines. Thermal, photostability, accelerated and real time stability testing was carried out with the marketed tablet formulation of the drug. An Inertsil  $^{\odot}$  C-18 (250 mm x 4.6 mm, 5µ) column was used to carry out the chromatographic analysis. The mobile phase composed of methanol-water (pH 2.5 with formic acid of the aqueous part) (35:65 %v/v) (flow rate 1.0 mL/min; Detection wavelength 254 nm). The drug was found to be extremely stable and there was no degradation under various stressor conditions. Excellent linearity was observed in the range of 0.05–5.0 µg/mL (r<sup>2</sup>= 0.9998). The limits of detection (**LOD**) and quantification (**LOQ**) were 0.0008 and 0.0024 µg/mL respectively. The proposed method gave good recovery of the drug in the tablet formulation as well (100.8 % in control and 98.53 - 99.73 % in the various stability samples).

Keywords: Tapentadol, stress degradation, chromatographic, stability indicating, validation

### INTRODUCTION

Tapentadol HC1 (TPL) ((-)3-[(1R,2R)-3-(dimethylamino-1-ethyl-2-methylpropyl]-phenol hydrochloride) (trade names: Nucynta<sup>®</sup> by Janssen Pharmaceuticals; Palexia®; Grünenthal Ltd. and Tapal<sup>®</sup> by MSN Labs) (Figure 1) is an oral centrally acting analgesic approved by the US FDA available as immediate release as well as controlled release formulations (tablets) for the treatment of moderate to severe pain in adults which is responsive only to opioid analgesics like oxycodone and morphine [1]. The drug is a unique opioid agonist which acts on µ opioid receptors and is also a noradrenaline reuptake inhibitor (NRI) [2-4]. The drug stability test guideline Q1A (R2) issued by the International Conference on

Harmonization [5] and WHO [6] suggest that inherent stability characteristics of a drug need to be studied under prescribed stress conditions. Tapentadol is not official in any pharmacopoeia and very few chromatographic methods are reported in literature for its analytical determination. The first reports included the determination of tapentadol and its metabolite N-desmethyltapentadol in oral and urine specimens by LC-MS<sup>n</sup> [7] and UPLC- MS<sup>n</sup> [8]. Subsequently, quantification of tapentadol was reported in canine plasma by Giorgim et al [9]. RP-HPLC methods for determination of tapentadol in bulk and pharmaceutical formulations have also been reported [10-11]. Spectrophotometric [12-13] and stability indicating spectrofluorometric methods [14] have been reported for determination of tapentadol in

bulk and pharmaceutical formulations. Two methods based on derivative spectrophotometry were recently reported for tapentadol [13,15]. Hence, the present study was designed to develop a stability indicating RP-HPLC method for the estimation of tapentadol HCl. The proposed method was validated with respect to various parameters outlined in the ICH guideline Q2(R1) (ICH, 2005) [16], i.e., specificity, linearity, accuracy, precision, limits of detection and quantification, system suitability parameters, ruggedness and robustness.

# MATERIALS AND METHODS

#### **Chemicals and Reagents**

Tapentadol HCl was graciously provided as gift sample by Ind-Swift Laboratories Ltd. N.A.C. Manimajra, Chandigarh. All chemicals and materials were of analytical grade and were purchased from Qualigens fine chemicals, Mumbai, India. Tapal® tablets (label amount 75 mg tapentadol HCl per tablet, MSN Laboratories Private Limited, Batch no. BT1410103A) were purchased from the market. Sodium hydroxide, hydrochloric acid and hydrogen peroxide (30%) were purchased from Loba chemical Pvt. Ltd (Mumbai, India). Methanol (MeOH), formic acid and acetonitrile (ACN) of HPLC grade were purchased from Merck Specialist Pvt Ltd, Mumbai (India). HPLC grade water obtained from Bio-Age Direct Ultra water purification system (Bio-Age Equipment & Services, Mohali, India) was used for preparation of all reagents and solutions.

## Instrumentation

#### LC–UV Analysis

The HPLC system consisted of binary pumps (515), UV detector and Rheodyne manual injector (Waters, Milfored, MA, USA) and the data was acquired and processed in Empower 3 software. An Inertsil® C-18 (250 mm x 4.6 mm, 5µ) column was employed for the chromatographic separation of the drug and degradation products. The mobile phase was filtered through nylon membrane (0.45µm) using Millipore filter assembly and was degassed using transonic sonicator bath (570/H ELMA, Germany). The chemicals were weighed on Afcoset electronic balance (ER-18 2A, Bombay Burmah trading Corp. Ltd., India). The pH of the buffer solution was adjusted using Digital pH meter (Sarthak, Panchkula, India). Laboratory centrifuge model CM 12 Plus (REMI) was employed for serum preparation and for processing of spiked serum samples.

### **Forced Degradation Studies**

The samples for hydrolytic and thermal stress testing were generated using high precision water bath and hot air oven equipped with digital temperature control capable of controlling temperature within range of  $\pm 1$  °C and  $\pm 2$  °C, respectively (Narang Scientific Works, New Delhi, India). Photodegradation of the drug was carried out in a photo stability chamber (Rolex Scientific Instruments, Ambala, India) capable of controlling temperature and humidity within range of  $\pm 2^{\circ}$  C and  $\pm 5$  % RH, respectively. The chamber was equipped with an illumination bank consisting of UV and fluorescent lamps as described in Option 2 made of light source as described in option 2 in the ICH guideline Q1B

[17]. The chamber was set at a temperature of 40°C

#### **Procedure for Forced Degradation Studies**

and 75% RH.

The drug was subjected to forced degradation under ICH prescribed stress conditions, viz., hydrolysis (acid, base and neutral), oxidation, photolysis and dry heat. Hydrolytic degradation was carried out under acidic, basic and neutral conditions by refluxing the drug in 0.1N HCl, 0.1N NaOH and water respectively (1 mg/mL) at 85 °C for 8 h. Photodegradation studies were carried out at 40 °C in a photostability chamber by exposing the solid drug in the form of a thin layer in a petri-dish and also drug solutions prepared in 0.1 N HCl, 0.1 N NaOH and H<sub>2</sub>O (0.1 mg/mL) to a total dose of 1.2 million lux h of fluorescent and 200 Wh/m<sup>2</sup> of UV-A illumination by placing them at about 9" from the light sources for 14 days. A parallel set of the drug solutions was stored in dark at the same temperature to serve as control. Oxidative degradation was carried out at room temperature in 30% hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) solution for 24 h using drug concentration of 1 mg/mL. Thermal degradation studies were carried out by exposing the drug (5 mg) sealed in amber coloured glass vials (5 mL), to a temperature of 50 °C for 31 days after which the vials were removed, cooled to room temperature and their contents dissolved in methanol.

#### **HPLC-UV** method

The UV absorption spectrum of the drug showed its absorption maximum ( $\lambda_{max}$ ) at 273 nm and hence, it was selected as detection wavelength for the HPLC method. The drug was chromatographed on C18 (250 mm x 4.0 mm; 5 $\mu$ , Inertsil<sup>®</sup>) column by a mobile phase composed of methanol and water (pH 2.5 with formic acid of aqueous part) (35:65 % v/v) at a flow rate of 1.0 mL/min. A Nucleosil<sup>®</sup> C8 (8mm×4.6mm i.d., 5  $\mu$ m) guard column was placed before the analytical column. The injection volume was 20  $\mu$ L, column temperature was the ambient temperature (30 ± 2 °C).

A stock solution of tapentadol HCl (1 mg/mL) was diluted upto 100 times with methanol to make a

standard solution (10  $\mu$ g/mL). For preparation of degradation samples for HPLC analysis, each degraded drug solution was diluted up to 100 times with methanol. The acid and alkali hydrolyzed solutions were mixed with equal volume of 0.1 N NaOH and 0.1 N HCl, respectively before diluting. For analysis of solid drug exposed to thermal and photolytic degradations, a 1 mg/mL solution of each sample was prepared and diluted up to 100 times with methanol. Each diluted sample was filtered through nylon membrane filter (0.45  $\mu$ , 13 mm) before injecting on HPLC.

### Validation of the method

The optimized method was validated with respect to various parameters outlined in the ICH guideline O2(R1) [16]. Linearity was assessed from serial dilutions (n=3) of the standard solution of the drug (10.0 µg/mL) in concentration range of 0.05-5.0 µg/mL (0.05, 0.1, 0.2, 0.5, 1.0, 2.0 and 5.0 µg/mL). Six injections, of three different concentrations (0.05, 0.2 and 2.0  $\mu$ g/mL), were given on the same day (to determine intra-day precision) and on three consecutive days (to determine inter-day precision). The precision was expressed as % RSD of each calculated concentration of the analyte. Method accuracy was evaluated as percent recovery (n=3) of three different concentrations of TPL (0.05, 0.5, and 2.0 µg/mL) from mixed degraded sample solutions  $(0.2 \ \mu g/mL)$  and was expressed as the percent recovery of the fortified drug concentration vis-à-vis the unfortified one. Robustness was assessed by carrying out deliberate changes in the method variables including change in detection wavelength, composition of mobile phase, pH of mobile phase, flow rate and column (Kromasil<sup>®</sup> and Inertsil<sup>®</sup>) one at a time and studying their impact on retention time (R<sub>T</sub>) and recovery of the drug in the test solution (n=3). Photodiode array detection method was used as an evidence for the specificity of method and to evaluate the homogeneity of the drug peak.

### **Stability Study of TPL Tablets**

The blister pack of TPL tablets (Tapal<sup>®</sup>, MSN Labs; label claim of 75 mg TPL per tablet) was exposed to accelerated conditions of 40 °C/75% RH in the photostability chamber for 6 months. Another blister pack was kept in the dark under similar conditions for 6 months. For thermal stability, the tablets were placed in the hot air oven maintained at 50 °C for 31 days. The real time stability samples were generated by keeping the tablets at room temperature ( $30 \pm 5$  °C) for 12 months. A control sample was kept in refrigerator at 4 °C. Each stability sample was analyzed by the validated HPLC method to quantify TPL concentration. The packed tablets from each stability condition were processed as follows: Ten tablets were weighed, crushed, powdered and a quantity of the powder equivalent to 10 mg of the drug was suspended in 8 mL of methanol in a 10 mL measuring flask. The contents were sonicated for 10 min, allowed to cool to room temperature and the volume was adjusted with methanol. The resultant stock solution was diluted with methanol to a drug concentration of 10  $\mu$ g/mL. The sample solutions were filtered through a 0.45  $\mu$ m membrane and analyzed (n=6) for the drug content using the validated HPLC method.

# **RESULTS AND DISCUSSION**

The LC-UV chromatograms of the standard solution of the drug and the various stress degraded solutions are shown in Figure 2. The drug was eluted as a symmetrical sharp peak at 7.80 min. under the employed chromatographic conditions.

### **Degradation behaviour**

Tapentadol HCl was found to be extremely stable under all the employed stressor conditions of acid/alkaline/neutral hydrolysis, oxidative, thermal and photolytic stress and no degradation products were found to be generated in any case.

#### **Method Validation**

The method was validated for parameters such as linearity, precision, accuracy, specificity and robustness and the results are listed in Table 1. A strictly linear relation was observed between the peak area and the concentration of TPL in the concentration range of 0.05-5.0  $\mu$ g/mL. The calibration plot (Figure 3) was described by the equation y = 275861.31 x + 748.20 (n = 3; r<sup>2</sup> = 0.9998).

The slopes and intercepts of calibration plots for three sets of peak areas in linearity studies were taken for calculation of LOD and LOQ values. Solutions of the drug with concentrations corresponding to LOD and LOQ values were prepared and analyzed six times (n = 6) and % RSD was calculated for the recovered amount determined from the corresponding calibration curve. The LOD and LOQ were 0.0008 and 0.0024 µg/mL with % RSD values of 1.81 % and 0.80 % respectively (Table 2). Peak areas in the accuracy studies with mixed degraded solutions of TPL showed good recoveries (98.35-99.96 %) of TPL at each fortification level with % RSD less than 1.17 % (Table 1). It suggested the method to be sufficiently accurate for the quantification of TPL.

The method was also found to be sufficiently precise with % RSD for the inter-day and intra-day precision less than 1.23 % and 0.62 % respectively (Table 1).

The method was found to be sufficiently robust as no significant changes in resolution, retention time, area and asymmetry of peak were observed after deliberate changes in the method variables including the detection wavelength, composition of mobile phase, pH of mobile phase, flow rate and column (Table 3). The % RSD for mean area corresponding to all variable conditions was found to be less than 0.3 %. The drug peak homogeneity (method specificity) was assessed by photodiode array detection and the purity angle for the drug peak was found to be less than the purity threshold, indicating

# Stability testing and assay on marketed formulation (Tapentadol HCl tablets)

the absence of any co-eluting peak.

The recovery results of the stability testing for TPL in the marketed formulation (tablet sample) under thermal, photostability, accelerated and real time stability conditions by the proposed method are shown in Table 4. Figure 4 shows the HPLC-UV chromatogram of the stability testing sample of tapentadol HCl (TPL) tablets. The percentage recovery in the control group was found to be 100.8 % (amount per tablet 75.6 mg) which was very close to the label claim (75 mg TPL per tablet). The percentage recovery in the various stability samples was found to be 98.53–99.73 % (amount per tablet **ISSN 2249-1848** 

found to be 73.9 - 74.8 mg) displaying a close agreement between the results obtained by the proposed methods and the label claim. This showed the suitability of the proposed method for assay of tapentadol in the marketed tablet formulation without any interference from the tablet excipients or routine degradation products.

#### CONCLUSION

In the present study, a simple, sensitive and reproducible stability-indicating HPLC-UV method was developed for determination of tapentadol hydrochloride. The method was validated for various parameters as per the ICH guidelines. Forced degradation study on tapentadol HCl indicated its stability to degradation under hydrolytic, thermal, oxidative and photolytic conditions. The proposed method was applicable to the determination of the drug in tablet formulation as well and the percentage recoveries were found to be 100.8 % in control and 98.53 - 99.73 % in the various stability samples.

#### Acknowledgements

We sincerely thank Ind-Swift Laboratories Ltd. N.A.C. Manimajra, Chandigarh for graciously providing us pure samples of tapentadol HCl.



Figure 1: Structure of tapentadol hydrochloride.



Figure 2: LC-UV chromatograms of standard solution of tapentadol (A) and the drug subjected to various degradation conditions, *viz.*, Hydrolysis in 0.1 M HCl (B); hydrolysis in 0.1N NaOH (C); hydrolysis in water (D); photolysis in 0.1N HCl (E); photolysis in 0.1N NaOH (F); photolysis in water (G); hydrolysis in 0.1N HCl in dark (H); hydrolysis in 0.1N NaOH in dark (I); hydrolysis in water in dark (J); Oxidation in H<sub>2</sub>O<sub>2</sub> (K); Thermal (L).







Figure 4:	HPLC-UV	<sup>7</sup> chromatogram	of stability	testing sami	ole of tai	pentadol (	TPL) tablets.
<b>.</b>							, , , , , , , , , , , , , , , , , , , ,

Parameter	TPL						
Accuracy	Concentration (µg/mL) ± S.D.; %RSD <sup>#</sup>						
	Spiked drug concn (µg/mL)*		Calculated**	%Recovery			
	0.05		$0.0492{\pm}\ 0.0006;\ 1.16\ \%$	98.35± 1.1444; 1.16 %			
	0.5		$0.4998{\pm}\ 0.0045;\ 0.89\ \%$	$99.96 {\pm}~ 0.8942 ; 0.89~\%$			
	2.0		$1.9910{\pm}\ 0.0134;\ 0.67\ \%$	99.55± 0.6706; 0.67 %			
Precision	Calculated concentration (µg/mL) ± S.D.; %RSD						
	Concn taken (µg/mL)		Intra-day $(n = 6)$	Inter-day $(n = 3)$			
	0.05		$0.0497 {\pm}~ 0.0003;  0.61\%$	$0.0500 \pm 0.0006; 1.22\%$			
	0.2 2.0		$0.2003 \pm 0.0006; 0.30\%$	$0.1982 \pm 0.0018; 0.89\%$			
			$1.9913 \pm 0.0064; 0.32\%$	$1.9988 \pm 0.0072; 0.36\%$			
Linearity	Range	Slope	Intercept	Coefficient of correlation r <sup>2</sup>			
	0.05-5.0 μg/mL	275861.31 (± 495.19)	748.20 (±66.63)	0.9998 (±0.00002)			
LOD	0.0008 µg/mL						
LOQ	0.0024 µg/mL						
Robustness	% RSD < 0.3 %						
Peak purity	Purity angle 0.244		Purity threshold				
			0.445	-			

Table 1: Validation parameters for the proposed method.

Interdier No	Peak areas for drug concentration					
Injection No. ——	0.0008 μg/mL*	0.0024 μg/mL**				
1	270	6543				
2	262	6402				
3	256	6478				
4	264	6432				
5	267	6489				
6	265	6512				
Mean	264	6476				
S.D.	4.7749	51.7107				
% RSD	1.81	0.80				

# Table 2. Recovery in LOD and LOQ studies.

\*LOD value

\*\*LOQ value

# Table 3. Robustness of the proposed method.

Parameter	Change	RT*	Peak Area		Mean	SD	%RSD	
Optimized conditions	NA	7.8	267932	268932	269032	268632	608.276	0.226
Flow Rate	0.7	8.12	268120	267988	268079	268062	67.560	0.025
	0.9	7.64	267956	268188	268579	268241	314.863	0.117
λ <sub>max</sub>	283	7.79	267750	267990	268402	268047	329.760	0.123
	263	7.82	267967	267998	268278	268081	171.310	0.064
Mobile Phase	30:70	8.2	267920	268288	268467	268225	278.889	0.104
Composition	40:60	7.72	268355	268079	268478	268304	204.331	0.076
Mobile Phase pH	2.4	7.77	267755	268088	268378	268073	311.747	0.116
	2.6	7.81	268250	267929	269089	268422	598.966	0.223
Column	Kromasil®	7.75	267944	268900	268480	268441	479.172	0.179
	Inertsil®	7.79	268356	267978	268099	268144	193.035	0.072

\*Retention time in minutes.

Stability condition	Mean recovery (mg) ± SD*; %RSD	Mean % recovery± SD <sup>#</sup> ; %RSD
Control (4°C)	75.6±0.42; 0.56%	100.8±0.56 %; 0.56%
Thermal (50°C; 31 days)	74.8±0.78; 1.04%	99.73±1.04 %; 1.04%
Photostability (40°C/75% RH, UV-VIS; 6 months)	74.8±0.60; 0.80%	99.73±0.80 %; 0.80%
Accelerated (40°C/75% RH; 4 months)	74.6±0.58; 0.78%	99.47±0.78 %; 0.78%
Real time (30 ± 5°C, 65 ± 5% RH; 12 months)	74.7±0.75; 1.00%	99.60±0.99 %; 1.00%

Table 4: Stability testing data of tapentadol tablets.

\*Calculated as mean of six measurements (n=6).

<sup>#</sup> Calculated as100xSD/mean.

#### REFERENCES

- 1. Tayal G, Grewal A, Mittal R, Bhatia N. J Anaes Clin Pharmacol, 2009; 25(4): 463-6.
- Tzschentke TM, Jahnel U, Kogel B, Englberger W, De Vry J, Schiene K, Okamoto A, Upmalis D, Weber H, Lange C, Stegmann JU, Kleinert R. Drugs Today (Barc), 2009; 45: 483–96.
- 3. Wade WE, Spruill WJ. Clin Ther, 2009; 31(12): 2804-18.
- 4. Sadeghi M, Tzschentke TM, Christie MJ. Br J Pharmacol, 2015; 172(2): 460-8.
- 5. ICH, Q1A (R2) Stability Testing of New Drug Substances and Products, International Conference on Harmonization, IFPMA, Geneva, 2003.
- 6. WHO, Draft stability testing of active pharmaceutical ingredients and pharmaceutical products, World Health Organization, Geneva, 2007.
- 7. Coulter C, Taruc M, Tuyay J, Moore C. J Anal Toxicol, 2010; 34(4): 458–63.
- 8. Bourland JA, Collins AA, Chester SA, Ramachandran S, Backer RC. J Anal Toxicol, 2010; 34: 450–7.
- 9. Giorgim M, Meizler A, Mills PC. J Pharm Biomed Anal, 2012; 67-68: 148-53.
- 10. Ramanaiah G, Ramachandran D, Srinivas G, Jayapal G, Rao P, Srilakshmi V. Int J Chem Anal Sci, 2012; 4(7): 391-6.
- 11. Gandhi J, Shah NJ, Lumbhani AN. Pharma Sci Monitor, 2012; 3(4): 2440-53.
- 12. Pavan AB, Mahesh J, Vijayalakshmi M. J Chem Pharm Sci, 2012; 5(2): 52-5.
- 13. Mobrouk MM, El-Fatatry HM, Hammad SF, Mohamed AA. J Appl Pharm Sci, 2013; 3(3): 122-5.
- 14. Panikumar DA, Haripriya A, Sirisha N, Raju YV, Sunitha G, Rao AV. J Appl Pharm, 2013; 5(3): 794-804.
- 15. Babu BS, Pavan KK, Nataraj K, Ramakrishna N. Der Pharm Lett, 2013; 5(2): 377-82.
- 16. ICH, Q2(R1), Validation of analytical procedures: Text and methodology, in: International Conference on Harmonisation, IFPMA, Geneva, 2005.
- 17. ICH, Q1B, Guidelines on photostability testing of new drug substances and products, in: Proceeding of International Conference on Harmonization, IFPMA, Geneva, 1996.